

Experimental and computational study on effect of ash deposition on fluid dynamic behavior in a bubbling fluidized bed gasifier

Rajan K Thapa Saroj Thapa Rajan Jaiswal Nora C I Furuvik Britt M E Moldestad

Department of Process Energy and Environmental Technology, University of Southeastern Norway, Norway,
{rajan.k.thapa, nora.c.i.furuvik, britt.moldestad}@usn.no
213843@student.usn.no, rajanjaiswal357@outlook.com

Abstract

The effect of ash deposition on fluid dynamic behavior in a fluidized bed gasification reactor have been studied using experimental and computational methods. The experiments were carried out using sand particles as bed material and air as fluidizing agent. A 3D computational model has been developed for a bubbling fluidized bed gasification reactor. First, the model was simulated using only sand particles and air. The results are compared with the experimental results. The comparison shows good agreement between the two sets of the results.

The model was further used to study the effect of ash accumulation on the fluid dynamic properties of a biomass gasification reactor. The bed material was mixed with 2 and 4vol% of ash and simulated in cold conditions. Pressure drop increases and minimum fluidization velocity decreases with increasing the ash deposition in the bed. The model was also simulated for 2, 4 and 6 vol% of ash at a temperature of 800°C. The minimum fluidization velocity was decreased in all the cases. The particle species concentration show the ash particles start to segregate at the minimum fluidization condition and are totally segregated at higher velocities. The bubble behavior of the bed is not effected by ash deposition.

Keywords: gasification, fluidized bed, segregation, mixing, CPFD, ash deposition

1 Introduction

Background

Gasification is the conversion of carbonaceous feedstock to the gaseous fuel by oxidation at a specific temperature. Gasification of biomass gives products such as carbon monoxide, hydrogen and methane. The mixture of these gases is called producer gas.

Fluidized bed biomass gasification is a high temperature process where the producer gas is the main product and ash is a byproduct. In some cases, the gasification temperature can reach as high as the ash melting temperature causing the ash to melt inside the bed. This creates bed agglomeration and consequently de-

fluidization of the reactor. Therefore, the ash melting phenomenon is undesirable in the gasification reactor.

Ash melting is also known as ash fusion or ash softening. This can also occur only in some of the hot spots in the bed. There are various reasons for the temperature to rise above the ash melting temperature. One of the reasons can be a high amount of ash deposition in the bed. Excessive ash deposition can reduce uniform mixing and heat transfer between bed material, biomass and fluidizing gas. The other reason can be the lower melting temperature of biomass ash which depend on its composition. Segregation between the deposited ash and bed material can also affect the mixing and heat transfer in the bed which can cause rapid increase in temperature in some areas of the bed.

Many researchers have studied bed agglomeration in biomass gasification reactors. Bartle's et al. have investigated the mechanism of agglomeration and concluded the fact that agglomeration takes place due to adhesiveness of bed material caused by alkali compounds present in the biomass. Pietsch (Pietsch, 2008) defined agglomeration as the formation of larger entities from particulate solid by sticking particles together by short range physical forces between the particles themselves, or through substances that adhere chemically or physically to the solid surface and form a material bridge between the particles. Visser and Tardos (Visser et al. 2008, Tardos et al. 1995) found two different routes for the initiation of bed agglomeration: melt-induced agglomeration and coating-induced agglomeration. Melt-induced means its adhesiveness with the bed particles that acts like glue. Coating-induced agglomeration implies the formation of coats between bed material and liquid phase due to adhesive force or stickiness. Tardos and Pfeffer (Tardos et al. 1995) found that the fluidization behavior is changed according to the agglomeration of bed particles. The bubbling behavior can be useful to indicate the agglomeration in BFB (Montes, 2014).

A number of bubbles passing though the bed in a specific interval of time is known as the bubbling frequency. Generally, the bubbling frequency is the same in all sections of the bed, but the bubbling frequency becomes different due to agglomeration. The

pressure drop through the bed is equal to the total hydrostatic pressure of the bed during fluidizing condition. So, agglomeration of the fluidized bed is characterized by lower pressure drop (Montes, 2014, Furuvi, et al. 2019).

Some of the major elements in biomass ash are K, Na, Ca, Si, P, Mg and Al. In order to study the behavior of ash deposition in relation to the fluid dynamic behavior, and develop a computational model for ash, it is important to study physical properties such as density, particle size, sphericity and melting temperatures of these elements and the percentage of the major elements in biomass ash. The study of the melting behavior and deposition of the ash are essential to make the biomass gasification reactor more efficient (Brown et al. 2005). Usually, high temperature is more favorable for carbon conversion because it produces less amount of tar. However, it brings other technological issues like ash melting and bed agglomeration (Niu et al. 2018). The alkali metals present in the biomass ash, will together with sulphur and chlorine in biomass fuels contribute to the bed agglomeration in fluidized bed reactors.

Biomass ash can be divided into three different groups based on their compositions. They are summarized in Table 1.

Table 1. Types of biomass ash

Biomass ash group	Ash components	Sintering Temperature
Woody biomass	Ca, K rich and Si lean	900-1000°C
Rice husk, bagasses or reed canary grass	Si-rich, Ca, K lean	700-1000°C
Sunflower seed and rapeseed seeds	Ca, K and P rich	700-1000°C

The ash melting takes place due to the low melting point of components such as K, Na and S that are present in woody biomass. The melted components start to bind together like glue and causes agglomeration of ash particles in the reactor (Vuthaluru, et al. 1999). Bed agglomeration eventually causes de-fluidization which reduces the efficiency and might lead to total shutdown of the bed. Moreover, ash deposition can cause corrosion in the reactor surface.

The studies are helpful for the prediction of ash melting and agglomeration of bed. The studies of the effect of the ash deposition on flow behavior in fluidized bed reactor can prevent, in some cases, rising of the bed temperature above the ash melting temperature. Therefore, this work is focused on the ash deposition from woody biomass in fluidized bed gasification reactors and the effect of the ash deposition on the fluid dynamic behavior in the reactor.

2 Experimental set up and procedure

The arrangements of the experimental setup is shown in Figure 1. The setup consists of a 1.4 m long, transparent column. The diameter of the column is 8.4 cm. Pressure transducers are connected to the tapping points, which are located along the height of the column. The tapping points are arranged along the column in different heights, and the distance between each tapping point is 10 cm. At the bottom of the column, there is an air distributor. Compressed air is supplied from the bottom of the column and the flow rate is measured by a flow-measuring meter. The flow of the compressed air is controlled by valves. The particles were filled from the top of the column.

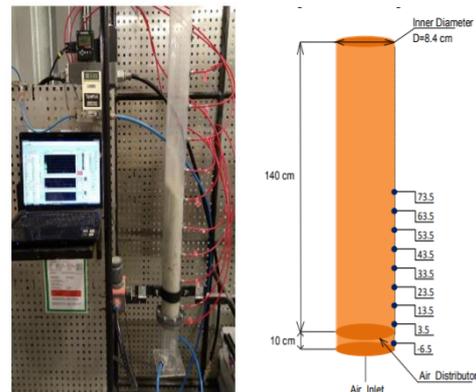


Figure 1. Experimental set up

The air flow rate and the pressure drop along the height of the column were logged in a LabView program. For each air velocity, data was recorded for more than 1 minute (sampling time 1 s), and a minimum of the 60s was allowed to establish the flow before the data was logged. Data Acquisition was done by Lab View and the data was imported in MATLAB and used to plot the pressure drop against superficial gas velocity. The pressure drop can be calculated at each tapping point by subtracting the distributor pressure at a similar flow rate.

$$\Delta P_{f1} = P_{f1} - D_{f1} \quad (1)$$

where, ΔP_{f1} is pressure reading at the pressure tapping point, D_{f1} is the distributor pressure and P_{f1} is the pressure at the pressure tapping point 1.

The properties of bed material used in the experiments are given in Table 2.

Table 2. Properties of bed materials and fluidizing gas

Particle mean diameter	285µm
Density	2650 kg/m ³
Bulk density	1428 kg/m ³
Solid volume fraction	0.54
Air density	1.25 kg/m ³

3 Computational model

A Computational Particle Fluid Dynamic (CPFD) model have been implemented to simulate the gas-solid flow with heat transfer. The commercial CPFD software Barracuda VR 17 is used for the simulations. The CPFD numerical methodology incorporates the multi-phase-particle-in-cell (MP-PIC) method. The gas phase is solved using Eulerian grid, and the particles are modeled as Lagrangian computational particles. Gas and particle momentum equations are solved in three dimensions. The fluid is described by the Navier-Stokes equation with strong coupling to the discrete particles. The particle momentum follows the MP-PIC description, which is a Lagrangian description of the particle motions described by ordinary differential equations with coupling with the fluid. More details about the computational methods are found in literature (Andrews, M et al.1996 and Snider D. 2001). The particles are grouped into parcels that contain a number of adjacent particles with similar properties such as density, size, species and velocity. The parcels are called computational particles. The computational particle is a numerical approximation similar to the numerical control volume where a spatial region has a single property for the fluid. With these computational particles, large commercial systems containing billions of particles can be simulated using millions of computational particles. This possibility of the CPFD numerical method is used in this work to simulate the ash deposition in a fluidized bed with sand particles as bed material.

3.1 Governing equations

The gas phase mass and momentum conservation are modeled with continuity and time averaged Navier-Stokes equations:

$$\frac{\partial(\alpha_g \rho_g)}{\partial t} + \nabla \cdot (\alpha_g \rho_g u_g) = 0 \quad (2)$$

$$\begin{aligned} \frac{\partial(\alpha_g \rho_g u_g)}{\partial t} + \nabla \cdot (\alpha_g \rho_g u_g u_g) \\ = -\nabla P + F + \nabla \cdot (\alpha_g \tau_g) \\ + \alpha_g \rho_g g \end{aligned} \quad (3)$$

where, α_g , ρ_g and u_g are gas phase volume fraction, density and velocity respectively. F is total momentum exchange with the particle phase per volume, g is gravitational acceleration, P is the pressure and τ_g is the gas phase stress tensor, which is given by:

$$\tau_g = \mu_g \left[(\nabla u_g + \Delta u_g^T) - \frac{2}{3} \nabla \cdot u_g I \right] \quad (4)$$

μ_g refers to the shear viscosity that is the sum of the laminar and turbulent components. The large eddy simulation is used for the large-scale turbulence

modeling while the subgrid scale turbulence is captured with Smagorinsky model:

$$\mu_{g,t} = C_s \rho_g \Delta^2 |\nabla u_g + \Delta u_g^T| \quad (5)$$

Where Δ is the subgrid length scale and calculated by equation 5. The default value for the model constant C_s is 0.01.

$$\Delta = (\delta x \delta y \delta z)^{1/3} \quad (6)$$

The interface momentum transfer is calculated through the viscous drag force:

$$F = \iint f \left\{ m_p \left[D_p(u_g - u_p) - \frac{\nabla P}{\rho_p} \right] \right\} dm_p du_p \quad (7)$$

Subscript p refers to the particle phase properties where m and u symbolize the mass and velocity. D_p is the drag function. The particle phase dynamics are derived using particle distribution function (PDF) calculated from the Liouville equation given as:

$$\frac{\partial f}{\partial t} + \nabla(f u_p) + \nabla u_p(f A_p) = 0 \quad (8)$$

where A_p , is the particle acceleration and is expressed by:

$$A_p = \frac{\partial(u_p)}{\partial t} = D_p(u_g - u_p) - \frac{\nabla P}{\rho_p} - \frac{\nabla \tau_p}{\rho_p \alpha_p} + g \quad (9)$$

α_p is particle volume fraction and τ_p is particle stress function that is used in formulating interphase interactions of particles.

$$\alpha_p = \iint f \frac{m_p}{\rho_p} dm_p du_p \quad (10)$$

$$\tau_p = \frac{10 P_s \alpha_p^\beta}{\max[(\alpha_{cp} - \alpha_p), \varepsilon(1 - \alpha_p)]} \quad (11)$$

P_s is a constant with the units of pressure, α_{cp} is the particle volume fraction at close packing, β is a constant between 2 and 5 where ε is a very small number on the order of 10⁻⁷.

3.2 Model parameters and geometry

A cylindrical CAD geometry with height of 1.4m, and diameter 8.4 cm was imported in Barracuda VR. The geometry is divided into 8652 uniform grids. After the grid generation, the base materials, the particle species, the pressure boundary condition and the flow boundary condition were specified in Barracuda which are similar to the experimental settings. The grid and boundary conditions are shown in Figure 2. No particle exit was considered during the simulation. The pressure monitoring points were selected 3.5 cm, 13.5 cm, and

23.5 cm along the height of the bed as in the experiments.

In Figure 2, the top of the cylinder yellow colored) is pressure outlet boundary and the bottom of the reactor (red colored) is velocity inlet boundary. The particles and gas are used similar to the experiments. The simulations were run in cold bed as well as in hot bed conditions to analyze the flow properties in real bed with various volume percent of ash deposition.

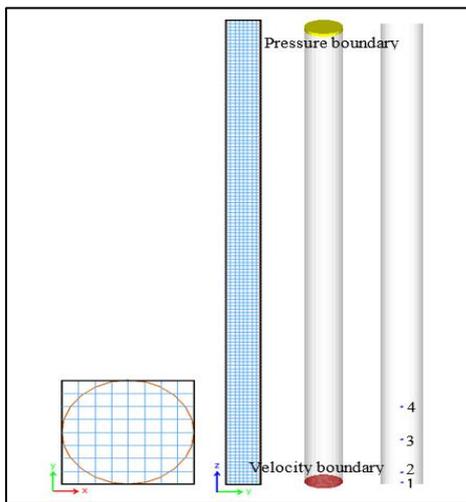


Figure 2. Grid and boundary conditions

The Wen-Yu drag model was used with 60% momentum loss after the particle collision. The initial height of the bed was 21 cm which gives the bed aspect ratio (ratio of height to diameter) of 2.5. Initial time step was taken as 0.001 s. Initial particle volume fraction was taken as 0.54. The simulations were performed for 120s of simulation time. Simulations were run for several air velocities and with different volume percent of ash mixed with sand particles to analyze the flow properties variation with the ash deposition.

Therefore, only the major elements of the ash composition were taken for the model. The composition and properties of ash particles used in the model are presented in Table 2.

Table 2. Properties biomass ash

SiO ₂	30%
K ₂ O	17%
CaO	53%
Average density	2959Kg/m ³
Molar mass	63.75 g/mol
Particle Size	50µm
Solid volume fraction	0.6
Sphericity	0.7

The minor components of the ash are neglected and the density and molar mass of the ash are calculated according to the densities and molar masses of the major components.

4 Results and discussion

4.1 Experimental results

A series of the experiments were carried out with different superficial air velocities and the pressure drop along the height of the bed were calculated using the experimental data. The plot of pressure drop as a function of superficial velocities (Figure 3) shows that the minimum fluidization velocity for sand particles is 0.075 m/s and the pressure drop at the minimum fluidization velocity is 13084 Pa/m.

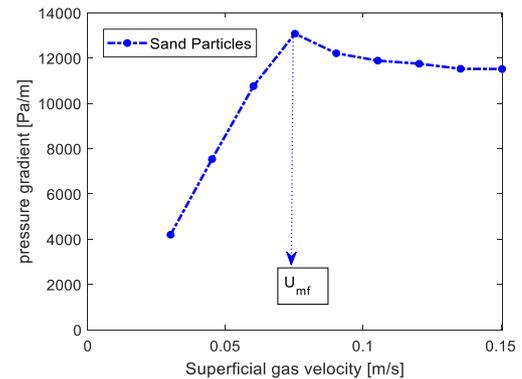


Figure 3. Pressure drop vs. superficial air velocity

It was difficult to run experiment with the mixture of sand particles and ash because the ash particles are very small and fine particles, which could leave the reactor. It was actually the main reason for developing the CPF model and simulating the flow of sand particles with ash.

4.2 Computational results

Simulations were run for the bed with sand particles, which is exactly similar to the experimental condition. The simulation results are compared against the experimental. Figure 4 shows comparison of experimental and computational pressure drops along the bed and minimum fluidization velocity of the bed.

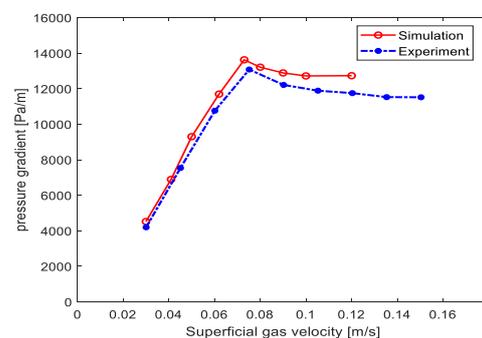


Figure 4. Comparison of pressure drop and minimum fluidization velocities

The figure shows good agreements between the experimental and computational results. The pressure drop and minimum fluidization velocities are almost similar in both cases. The aim of this comparison was to validate a CPF model for its further use for simulation

of a bed with various volume percent of ash deposition. Therefore, a series of simulation were run to simulate the bed of sand particles with 2% and 4vol% of ash particles by volume. The simulation were run for cold conditions at room temperature.

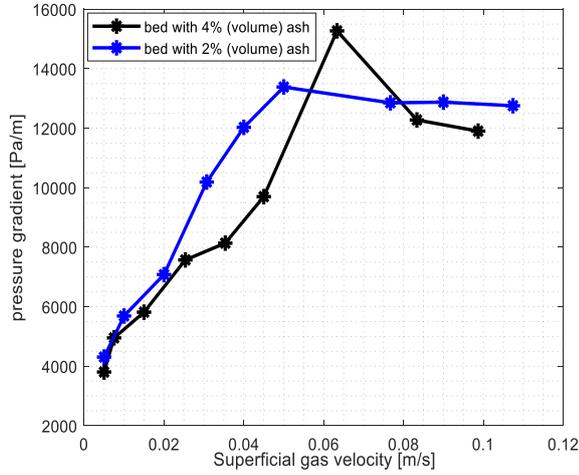


Figure 5. Minimum fluidization velocities for the mixture of ash with sand.

The pressure drop is increased with increasing ash volume percentage. The minimum fluidization velocities decreased with increasing volume percent of ash as shown in Figure 5. Minimum fluidization velocity for beds of small particles is strongly depended on bed void fraction [13], which can be seen from the equation:

$$u_{mf} = \frac{d_p^2(\rho_p - \rho_g)g}{150\mu} \cdot \frac{\varepsilon_{mf}^3 \phi_s^2}{1 - \varepsilon_{mf}} \quad (12)$$

where, d_p is particle diameter, ρ_p and ρ_g are particle and gas densities, ε_{mf} is bed void fraction, ϕ_s is sphericity and μ is gas viscosity. Presence of ash particles which have small particle size compared to the sand having larger particles size makes ash particles to fill the void between the sand particles reducing the void fraction in the bed. This is the reason for the reduction of minimum fluidization velocity. This means a limited ash deposition in the bed does not have negative impact in flow properties (particularly in fluidization velocity) before the sand particles are segregated in the bed.

Simulations were also run at high temperature. Figure 6 shows the simulation results at the temperature of 800°C. The bed was mixed with sand particles and 2%, 4% and 6% by volume of ash particles. The results show that an increase in the temperature decreases the minimum fluidization velocities. When the temperature of fluidizing air is increased from 20°C to 800°C, the viscosity of air increases from $1.825 \cdot 10^{-5} \text{Pa} \cdot \text{s}$ to $4.362 \cdot 10^{-5} \text{Pa} \cdot \text{s}$. The increase in viscosity is the reason for the decrease in minimum fluidization velocity (See equation 12). The results show that the ash deposition up to 6% by volume has no negative effect on the flow behavior in the reactor. However the pressure drop in the bed increases with increasing ash

deposition. The deposition will have negative effect only when the temperature of any spot of the bed reaches the ash melting temperature. In order to avoid this, it is important to enhance mixing of particles in bed and avoid segregation.

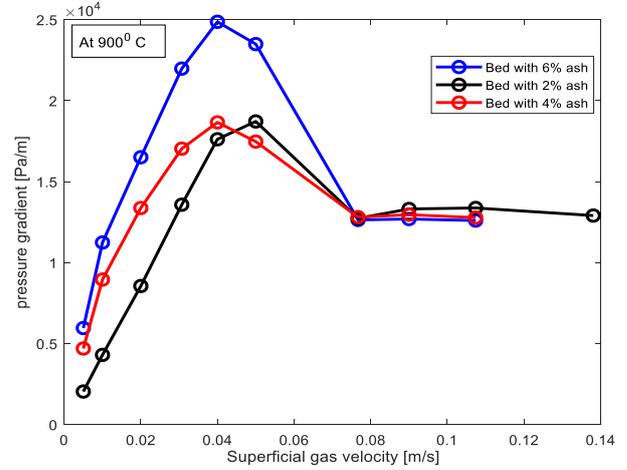


Figure 6. Minimum fluidization velocity at high temperature conditions.

The CPFD model allows tracking the particles in the bed separately in the form of different particle species. The distribution of ash and sand particles in the bed are shown in Figure 7. The figure shows the comparison of species distribution in the bed at the minimum fluidization velocity and the air velocity of 0.21 m/s for the bed with 2, 4 and 6 vol% ash respectively from the left to the right.

The figure shows that a part of the ash particles have segregated and moved to the top of the bed already at the minimum fluidization condition and the segregation becomes more distinct at the higher air velocities.

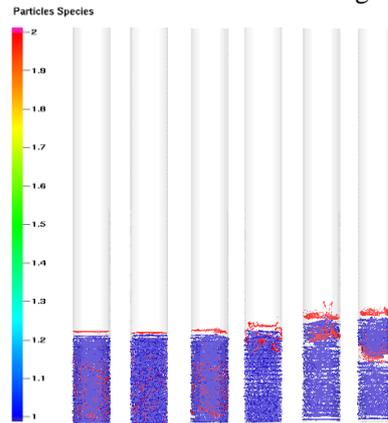


Figure 7. Sand and ash particle species distribution in the bed

Deposited ash, after segregation can effect negatively on the mixing and heat transfer in the bed. This is more probable at the top of the bed where only ash particles are present. This can create hot spots at the upper part of the bed.

The bubble behavior in the bed at the air velocity of 0.21 m/s are shown in Figure 8. The figure shows that there is no significant effect of the ash deposition on the

